

Fig. 1. View of $\left[\mathrm{Pt}(\mathrm{bpy}) \mathrm{Cl}_{4}\right]$ with the atom-numbering scheme. Thermal ellipsoids are drawn at the $50 \%$ probability level.
the equivalent $\mathrm{Pd}-\mathrm{Cl}$ distances though the same relative trans effects on these distances are observed in each structure.

Least-squares-planes' calculations show that the two rings of the bipyridyl group are essentially planar |all deviations less than 0.04 (2) $\AA$ I, are coplanar, meeting at an angle of $2 \cdot 2(7)^{\circ}$, and that no atoms of the ligand deviate by more than 0.08 (2) $\AA$ from the plane defined by $\mathrm{Pt}, \mathrm{Cl}(3), \mathrm{Cl}(4), \mathrm{N}(1)$ and $\mathrm{N}(2)$.

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# A 1:1 Complex between 1,4,7,10,13,16-Hexaoxacyclooctadecane (18-Crown-6) and Mercury(II) Iodide 

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#### Abstract

C}_{12} \mathrm{H}_{24} \mathrm{O}_{6} \cdot \mathrm{HgI}_{2}, \quad M_{r}=718 \cdot 7\), monoclinic, $C 2 / m, a=13.049$ (5), $b=11.241$ (3), $c=9.285$ (2) $\AA$, $\beta=134.89(3)^{\circ}, \quad V=965 \AA^{3}, \quad Z=2, \quad D_{x}=$ $2.47 \mathrm{~g} \mathrm{~cm}^{-3}, \quad$ Cu $K \alpha, \quad \lambda=1.54178 \AA, \quad \mu=410 \mathrm{~cm}^{-1}$, $F(000)=660$, room temperature, $R=0.036$ for 558 unique observed reflections with $\left|F_{o}\right|>3 \sigma\left(\left|F_{o}\right|\right)$. The complex has crystallographic $2 / m$ symmetry and approximate molecular $D_{3 d}$ symmetry. The structure is composed of linear $\mathrm{HgI}_{2}$ entities inserted normal to the mean plane of the macrocycle with the Hg atom positioned at its centre. The coordination of the Hg atom is hexagonal bipyramidal with axial $\mathrm{Hg}-\mathrm{I}$ bond distances of 2.622 (1) $\AA$ and a mean equatorial $\mathrm{Hg} \cdots \mathrm{O}$ distance of 2.86 (2) $\AA$. The accommodation of the large Hg atom within the macrocycle results in a significant


increase in the magnitude of the $\mathrm{O}-\mathrm{C}-\mathrm{C}-\mathrm{O}$ torsional angles from the values normally observed in complexes of 18 -crown- 6 in which the ligands adopt the all-gauche conformation with pseudo $D_{3 d}$ symmetry.

Introduction. During our recent investigations (Allwood, Crosby, Pears, Stoddart \& Williams, 1984) on the complexation of sulfonium cations by 18 -crown-6 (18C6), we attempted to isolate a $2: 1$ crystalline complex between $\mathrm{MeSPh}_{2}^{+} . \mathrm{HgI}_{3}^{-}$and 18 C 6 from MeOH . In fact, the first crystals we isolated corresponded to a $1: 1$ complex between $\mathrm{HgI}_{2}$ and 18C6, indicating that $\mathrm{MeSPh}_{2}^{+} . \mathrm{HgI}_{3}^{-}$had disproportionated to $\mathrm{MeSPh}_{2}^{+} . \mathrm{I}^{-}$and $\mathrm{HgI}_{2}$ under the conditions of the crystal-growing experiment. In view of the
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interest which is currently being shown in the solid-state structures of $\left[\mathrm{CdCl}_{2} .18 \mathrm{C} 6\right]$ and $\left[\mathrm{HgCl}_{2} .18 \mathrm{C} 6\right]$ (Paige \& Richardson, 1984) and their dibenzo-18-crown-6 (DB18C6) analogues (Kawasaki \& Matsuura, 1984), in addition to the previously described (Malmsten, 1979) crystalline $1: 1$ complexes of $\mathrm{CdI}_{2}$ and $\mathrm{HgI}_{2}$ with 1,4,10,13-tetraoxa-7,16-diazacyclooctadecane (DA18C6), we now report on the fortuitous isolation and crystal structure of [ $\left.\mathrm{HgI}_{2} \cdot 18 \mathrm{C} 6\right]$.

Experimental. A solution of $\mathrm{MeSPh}_{2}^{+} \cdot \mathrm{HgI}_{3}^{-}(200 \mathrm{mg})$ in warm MeOH ( 15 ml ) was added to a solution of 18 C 6 ( 34 mg ) in $\mathrm{MeOH}(5 \mathrm{ml})$. The solution was allowed to stand at room temperature for 2 h and the resulting crystals initially isolated were found to be suitable for X-ray structural investigation. Crystal size $0.05 \times$ $0.05 \times 0.05 \mathrm{~mm}$. Refined unit-cell parameters obtained by centring 18 reflections. Nicolet $R 3 m$ diffractometer. 721 independent reflections $\left(\theta \leq 50^{\circ}\right)$ measured, $\mathrm{Cu} K \alpha$ radiation (graphite monochromator), $\omega$ scan; $558\left[\left|F_{o}\right|>3 \sigma\left(\left|F_{o}\right|\right)\right]$ considered observed, index range $h-14 / 10, k 0 / 12, l 0 / 10$; two check reflections measured every 50 reflections, net count constant; Lorentz and polarization corrections, numerical absorption correction (max. and min. transmission factors, 0.30 and $0 \cdot 10$ ) for face-indexed crystal. Structure solved by heavy-atom method; non-hydrogen atoms refined anisotropically; positions of H atoms calculated ( $\mathrm{C}-\mathrm{H} \quad 0.96 \AA$ ); H atoms assigned isotropic thermal parameters, $U(\mathrm{H})=$ $1 \cdot 2 U_{\text {eq }}(\mathrm{C})$, and allowed to ride on parent C atoms. An empirical extinction correction was applied $[g=$ 0.0012 (2)]. Refinement using $F$ magnitudes by blockcascade full-matrix least squares; $R=0.036 ; w R$ $=0.039\left[w^{-1}=\sigma^{2}(F)+0.0003 F^{2}\right] ;(\Delta / \sigma)_{\max }=0.002$; residual electron density in difference map within -1.0 and +0.82 e $\AA^{-3}$; atomic scattering factors and dispersion corrections from International Tables for X-ray Crystallography (1974). Computations carried out on an Eclipse S140 computer using the SHELXTL program system (Sheldrick, 1983).

Discussion. Table 1* lists the fractional atomic coordinates of the non-hydrogen atoms, Table 2 the bond lengths and angles and the $\mathrm{C}-\mathrm{C}$ and $\mathrm{C}-\mathrm{O}$ torsional angles in 18C6.

The structure of [ $\mathrm{HgI}_{2} .18 \mathrm{C} 6$ ] is illustrated in Fig. 1, which also shows the atomic numbering scheme. Fig. $2 \dagger$ shows a space-filling representation of the complex.

[^0]There are several examples of 18C6 (Paige \& Richardson, 1984; Kawasaki \& Matsuura, 1984), DA18C6 (Malmsten, 1979), and DB18C6 (Henrick, Matthews, Podejma \& Tasker, 1982; Crowder, Henrick, Matthews \& Podejma, 1983), and of the cis-cisoid-cis and cis-transoid-cis isomers of dicyclo-hexano-18-crown-6 (Hughes \& Truter, 1983) forming complexes with guest molecules such as $\mathrm{CdCl}_{2}, \mathrm{HgCl}_{2}$, $\mathrm{HgI}_{2}$, and $\mathrm{TlMe}_{2}^{+}$, threaded through the centre of the

Table 1. Atom coordinates ( $\times 10^{4}$ ) and temperature factors $\left(\AA^{2} \times 10^{3}\right)$ with e.s.d.'s in parentheses

|  | $x$ | $y$ | $z$ | $U_{\mathrm{eq}}{ }^{*}$ |
| :--- | :---: | :---: | :---: | :---: |
| Hg | 0 | 0 | 0 | $39(1)$ |
| I | $-2758(1)$ | 0 | $-3399(2)$ | $55(1)$ |
| $\mathrm{O}(1)$ | $-486(13)$ | 0 | $2572(18)$ | $47(9)$ |
| $\mathrm{C}(2)$ | $-1247(14)$ | $1057(13)$ | $2204(20)$ | $56(11)$ |
| $\mathrm{C}(3)$ | $-435(17)$ | $2101(14)$ | $2501(21)$ | $63(12)$ |
| $\mathrm{O}(4)$ | $-632(9)$ | $2195(7)$ | $793(13)$ | $51(7)$ |
| $\mathrm{C}(5)$ | $146(17)$ | $3138(12)$ | $957(24)$ | $66(12)$ |

[^1]Table 2. Bond lengths $(\AA)$, bond angles $\left({ }^{\circ}\right)$, and torsion angles $\left(^{\circ}\right)$

| $\mathrm{Hg}-\mathrm{O}(1)$ | $2.871(21)$ | $\mathrm{Hg}-\mathrm{O}(4)$ | $2.858(10)$ |
| :--- | :--- | :--- | ---: |
| $\mathrm{Hg}-\mathrm{I}$ | $2.622(1)$ | $\mathrm{O}(4)-\mathrm{C}(5)$ | $1.401(22)$ |
| $\mathrm{O}(1)-\mathrm{C}(2)$ | $1.426(19)$ | $\mathrm{C}(3)-\mathrm{O}(4)$ | $1.420(27)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.471(25)$ | $\mathrm{C}(5)-\mathrm{C}\left(5^{\prime}\right)$ | $1.531(50)$ |
|  |  |  |  |
| $\mathrm{O}(4)-\mathrm{Hg}-\mathrm{O}\left(4^{\prime \prime}\right)$ | $119.4(6)$ | $\mathrm{O}(4)-\mathrm{Hg}-\mathrm{O}\left(4^{\prime}\right)$ | $60.6(6)$ |
| $\mathrm{O}(1)-\mathrm{Hg}-\mathrm{O}\left(4^{\prime}\right)$ | $119.2(3)$ | $\mathrm{I}-\mathrm{Hg}-\mathrm{O}\left(1^{\prime}\right)$ | $85.4(2)$ |
| $\mathrm{O}(1)-\mathrm{Hg}-\mathrm{O}(4)$ | $60.8(3)$ | $\mathrm{I}-\mathrm{Hg}-\mathrm{O}\left(4^{\prime}\right)$ | $95.2(1)$ |
| $\mathrm{I}-\mathrm{Hg}-\mathrm{O}(1)$ | $94.6(2)$ | $\mathrm{I}-\mathrm{Hg}-\mathrm{O}(4)$ | $84.8(1)$ |
| $\mathrm{I}-\mathrm{Hg}-\mathrm{I}^{\prime}$ | 180.0 | $\mathrm{C}(2)-\mathrm{O}(1)-\mathrm{C}\left(2^{11}\right)$ | $112.9(19)$ |
| $\mathrm{O}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $109.5(16)$ | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(4)$ | $108.1(12)$ |
| $\mathrm{C}(3)-\mathrm{O}(4)-\mathrm{C}(5)$ | $112.9(12)$ | $\mathrm{O}(4)-\mathrm{C}(5)-\mathrm{C}\left(5^{1}\right)$ | $107.8(9)$ |


| $\mathrm{C}\left(2^{1}\right)-\mathrm{O}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $-176.7(10)$ |
| :--- | ---: |
| $\mathrm{O}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(4)$ | $75.4(15)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(4)-\mathrm{C}(5)$ | $-177.6(13)$ |
| $\mathrm{C}(3)-\mathrm{O}(4)-\mathrm{C}(5)-\mathrm{C}\left(5^{\prime}\right)$ | $179.6(14)$ |
| $\mathrm{O}(4)-\mathrm{C}(5)-\mathrm{C}\left(5^{1}\right)-\mathrm{O}\left(4^{1}\right)$ | $-74.6(15)$ |

Symmetry code: (i) $-x, y,-z$; (ii) $x,-y, z$.


Fig. 1. Side view of the $1: 1$ complex and the atomic numbering scheme.


Fig. 2. Space-filling representation of $\left[\mathrm{HgI}_{2} \cdot 18 \mathrm{C} 6\right]$.
macrocyclic ring, i.e. they have rotoxane-like structures. Of particular note are the closely related 18C6 complexes (Paige \& Richardson, 1984; Kawasaki \& Matsuura, 1984) with $\mathrm{CdCl}_{2}$ and $\mathrm{HgCl}_{2}$, and the DA18C6 complex (Malmsten, 1979) with $\mathrm{HgI}_{2}$, where the gross structural features of the $1: 1$ complexes are analogous to that of [ $\left.\mathrm{HgI}_{2} \cdot 18 \mathrm{C} 6\right]$. The present structure provides another example of the relatively rare hexagonal-bipyramidal geometry. The $\mathrm{Hg}-\mathrm{I}$ bond distances $[2.622(1) \AA]$ are the same as those observed by Jeffrey \& Vlasse (1967) in $\mathrm{HgI}_{2}$, but significantly shorter than the value of 2.680 (1) $\AA$ reported by Malmsten (1979) for [ $\mathrm{HgI}_{2}$.DA18C6]. The $\mathrm{Hg} \cdots \mathrm{O}$ distances and the $\mathrm{I}-\mathrm{Hg} \cdots \mathrm{O}$ angles (Table 2) do not depart significantly from those reported (Paige \& Richardson, 1984) for [ $\mathrm{HgCl}_{2}$. 18C6]. A consequence of the accommodation of a large Hg atom at the centre of the 18C6 macrocycle is an increase in the magnitude of the mean $\mathrm{O}-\mathrm{C}-\mathrm{C}-\mathrm{O}$ torsional angle [75(2) ${ }^{\circ}$, $c f$. the 'normal' value of $\mathrm{ca} 65^{\circ}$ (Goldberg, 1980).

Surprisingly, although the title complex as a whole has within statistical significance $\overline{3}$ symmetry and
adopts a very similar packing arrangement to that reported by Paige \& Richardson (1984) for [ $\left.\mathrm{HgCl}_{2} \cdot 18 \mathrm{C} 6\right]$ and $\left[\mathrm{CdCl}_{2} .18 \mathrm{C} 6\right]$, the crystals are not rhombohedral. This may be a consequence of accommodating covalently larger I atoms in the structure in place of Cl atoms. In fact, the unit-cell parameters initially chosen by Paige \& Richardson (1984) for a $C$-face-centred cell of $a=10.44, \quad b=11.46, \quad c=$ $7.75 \AA$, and $\beta=82.21^{\circ}$ are all different from the present $C$-face-centred cell.

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# Structures of Dibromo[ $N, N^{\prime}$-bis(2-pyridylmethylene)-1,3-propanediamine]copper(II) (1) and Bromo[ $N, N^{\prime}$-bis(2-pyridylmethylene)-1,4-butanediamine]copper(II) Bromide (2) 

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$$
\begin{aligned}
& \text { Abstract. (1): }\left[\mathrm{CuBr}_{2}\left(\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{4}\right)\right], M_{r}=475 \cdot 7, \text { mono- } \\
& \text { clinic, } \quad C 2 / c, \quad a=13.286(3), \quad b=9.723(3), \quad c= \\
& 13.234(2) \AA, \quad \beta=107.54(1)^{\circ}, \quad V=1630 \cdot 1(5) \AA^{3}, Z \\
& =4, \quad D_{m}=1.94(3), \quad D_{x}=1.94 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda(\mathrm{Mo} K \alpha)= \\
& 0108-2701 / 86 / 010053-04 \$ 01.50
\end{aligned}
$$

[^2]
[^0]:    * Lists of structure factors, anisotropic thermal parameters, H -atom parameters and a space-filling diagram have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42495 ( 7 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.
    $\dagger$ A further space-filling diagram, highlighting the equatorial coordination of Hg by the six O atoms, has been deposited.

[^1]:    * $U_{\text {eq }}$ is defined as one third of the trace of the orthogonalized $U_{i j}$ tensor.

[^2]:    $0.7107 \AA, \mu=65.8 \mathrm{~cm}^{-1}, \quad F(000)=932, T=293 \mathrm{~K}$, $R=0.031$ for 1238 observed $[I>2 \sigma(I)]$ of 1898 unique reflections. (2): $\left[\mathrm{CuBr}\left(\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{4}\right)\right] \mathrm{Br}, M_{r}=$ 489.8, monoclinic, $P 2_{1} / c, \quad a=12.142(3), \quad b=$
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