

Fig. 1. View of $[Pt(bpy)Cl_4]$ with the atom-numbering scheme. Thermal ellipsoids are drawn at the 50% probability level.

the equivalent Pd-Cl distances though the same relative *trans* effects on these distances are observed in each structure.

Least-squares-planes' calculations show that the two rings of the bipyridyl group are essentially planar [all deviations less than 0.04 (2) Å], are coplanar, meeting at an angle of 2.2 (7)°, and that no atoms of the ligand deviate by more than 0.08 (2) Å from the plane defined by Pt, Cl(3), Cl(4), N(1) and N(2).

References

- BUSE, K. D., KELLER, H. J. & PRITZKOW, H. (1977). *Inorg. Chem.* **16**, 1072–1076.
- GRAY, L. R., GULLIVER, D. J., LEVASON, W. & WEBSTER, M. (1983). J. Chem. Soc. Dalton Trans. pp. 133-141.
- International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)
- JOHNSON, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
- OSBORN, R. S. & ROGERS, D. (1974). J. Chem. Soc. Dalton Trans. pp. 1002-1004.
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England.

Acta Cryst. (1986). C42, 51-53

A 1:1 Complex between 1,4,7,10,13,16-Hexaoxacyclooctadecane (18-Crown-6) and Mercury(II) Iodide

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(Received 1 April 1985; accepted 10 September 1985)

Abstract. $C_{12}H_{24}O_6 HgI_2$, $M_r = 718 \cdot 7$, monoclinic, C2/m, a = 13.049 (5), b = 11.241 (3), c = 9.285 (2) Å, $\beta = 134.89 \ (3)^{\circ}$, $V = 965 \text{ Å}^3$, $D_{r} =$ Z = 2,2.47 g cm⁻³, Cu Ka, $\lambda = 1.54178$ Å, $\mu = 410$ cm⁻¹, F(000) = 660, room temperature, R = 0.036 for 558 unique observed reflections with $|F_a| > 3\sigma(|F_a|)$. The complex has crystallographic 2/m symmetry and approximate molecular D_{3d} symmetry. The structure is composed of linear HgI₂ entities inserted normal to the mean plane of the macrocycle with the Hg atom positioned at its centre. The coordination of the Hg atom is hexagonal bipyramidal with axial Hg-I bond distances of 2.622 (1) Å and a mean equatorial Hg...O distance of 2.86 (2) Å. The accommodation of the large Hg atom within the macrocycle results in a significant

increase in the magnitude of the O–C–C–O torsional angles from the values normally observed in complexes of 18-crown-6 in which the ligands adopt the all-gauche conformation with pseudo D_{3d} symmetry.

Introduction. During our recent investigations (Allwood, Crosby, Pears, Stoddart & Williams, 1984) on the complexation of sulfonium cations by 18-crown-6 (18C6), we attempted to isolate a 2:1 crystalline complex between MeSPh₂⁺.HgI₃⁻ and 18C6 from MeOH. In fact, the first crystals we isolated corresponded to a 1:1 complex between HgI₂ and 18C6, indicating that MeSPh₂⁺.HgI₃⁻ had disproportionated to MeSPh₂⁺.HgI₃⁻ had disproportionated to MeSPh₂⁺.I⁻ and HgI₂ under the conditions of the crystal-growing experiment. In view of the

0108-2701/86/010051-03\$01.50

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interest which is currently being shown in the solid-state structures of [CdCl₂.18C6] and [HgCl₂.18C6] (Paige & Richardson, 1984) and their dibenzo-18-crown-6 (DB18C6) analogues (Kawasaki & Matsuura, 1984), in addition to the previously described (Malmsten, 1979) crystalline 1:1 complexes of CdI, and HgI, with 1,4,10,13-tetraoxa-7,16-diazacyclooctadecane (DA18C6), we now report on the fortuitous isolation and crystal structure of [HgI₂.18C6].

Experimental. A solution of MeSPh⁺₂.HgI⁻₃ (200 mg) in warm MeOH (15 ml) was added to a solution of 18C6 (34 mg) in MeOH (5 ml). The solution was allowed to stand at room temperature for 2 h and the resulting crystals initially isolated were found to be suitable for X-ray structural investigation. Crystal size $0.05 \times$ 0.05×0.05 mm. Refined unit-cell parameters obtained by centring 18 reflections. Nicolet R3m diffractometer. 721 independent reflections ($\theta \le 50^\circ$) measured, Cu Ka radiation (graphite monochromator), ω scan; 558 $[|F_o| > 3σ(|F_o|)]$ considered observed, index range h-14/10, k0/12, l0/10; two check reflections measured every 50 reflections, net count constant; Lorentz and polarization corrections, numerical absorption correction (max. and min. transmission factors, 0.30 and 0.10) for face-indexed crystal. Structure solved by heavy-atom method; non-hydrogen atoms refined anisotropically; positions of H atoms calculated (C-H 0.96 Å); H atoms assigned isotropic thermal parameters, U(H) = $1 \cdot 2U_{eq}(C)$, and allowed to ride on parent C atoms. An empirical extinction correction was applied [g =0.0012 (2)]. Refinement using F magnitudes by blockcascade full-matrix least squares; R = 0.036; wR = 0.039 $[w^{-1} = \sigma^2(F) + 0.0003F^2]; (\Delta/\sigma)_{max} = 0.002;$ residual electron density in difference map within -1.0and +0.82 e Å-3; atomic scattering factors and dispersion corrections from International Tables for X-ray Crystallography (1974). Computations carried out on an Eclipse S140 computer using the SHELXTL program system (Sheldrick, 1983).

Discussion. Table 1* lists the fractional atomic coordinates of the non-hydrogen atoms, Table 2 the bond lengths and angles and the C-C and C-O torsional angles in 18C6.

The structure of [HgI₂.18C6] is illustrated in Fig. 1, which also shows the atomic numbering scheme. Fig. 2[†] shows a space-filling representation of the complex.

There are several examples of 18C6 (Paige & Richardson, 1984; Kawasaki & Matsuura, 1984), DA18C6 (Malmsten, 1979), and DB18C6 (Henrick, Matthews, Podejma & Tasker, 1982; Crowder, Henrick, Matthews & Podejma, 1983), and of the ciscisoid-cis and cis-transoid-cis isomers of dicyclohexano-18-crown-6 (Hughes & Truter, 1983) forming complexes with guest molecules such as CdCl₂, HgCl₂, HgI₂, and TlMe⁺₂, threaded through the centre of the

Table	1.	Atom	coordinate	s (×10	14) an	d temper	rature
f	àct	ors (Ų	$\times 10^3$) with	e.s.d.'s	in pa	rentheses	

	x	у	Z	U_{eq}^*
Hg	0	0	0	39(1)
Ι	-2758 (1)	0	-3399 (2)	55 (1)
0(1)	-486 (13)	0	2572 (18)	47 (9)
C(2)	-1247 (14)	1057 (13)	2204 (20)	56 (11)
C(3)	-435 (17)	2101 (14)	2501 (21)	63 (12)
O(4)	-632 (9)	2195 (7)	793 (13)	51 (7)
C(5)	146 (17)	3138 (12)	957 (24)	66 (12)

* U_{eq} is defined as one third of the trace of the orthogonalized U_{ii} tensor.

Table 2. Bond lengths (Å), bond angles (°), and torsion angles (°)

HgO(1) Hg-I O(1)-C(2) C(2)-C(3)	2·871 (21) 2·622 (1) 1·426 (19) 1·471 (25)	Hg-O(4) O(4)-C(5) C(3)-O(4) C(5)-C(5')	2·858 (10) 1·401 (22) 1·420 (27) 1·531 (50)
$\begin{array}{l} O(4)-Hg-O(4^{ii})\\ O(1)-Hg-O(4^{i})\\ O(1)-Hg-O(4)\\ I-Hg-O(1)\\ I-Hg-I'\\ O(1)-C(2)-C(3)\\ C(3)-O(4)-C(5) \end{array}$	119.4 (6) 119.2 (3) 60.8 (3) 94.6 (2) 180.0 109.5 (16) 112.9 (12)	$\begin{array}{l} O(4)-Hg-O(4^{1})\\ I-Hg-O(1^{1})\\ I-Hg-O(4^{1})\\ I-Hg-O(4)\\ C(2)-O(1)-C(2^{11})\\ C(2)-C(3)-O(4)\\ O(4)-C(5)-C(5^{1}) \end{array}$	60.6 (6) 85.4 (2) 95.2 (1) 84.8 (1) 112.9 (19) 108.1 (12) 107.8 (9)
$\begin{array}{c} C(2^{i}) - O(1) - C(2) - C \\ O(1) - C(2) - C(3) - O(i) \\ C(2) - C(3) - O(4) - C(i) \\ C(3) - O(4) - C(5) - C(i) \\ O(4) - C(5) - C(5^{i}) - O(i) \\ \end{array}$	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	7 (10) 4 (15) 5 (13) 5 (14) 5 (15)	

Symmetry code: (i) -x, y, -z; (ii) x, -y, z.



Fig. 1. Side view of the 1:1 complex and the atomic numbering scheme.

^{*} Lists of structure factors, anisotropic thermal parameters, H-atom parameters and a space-filling diagram have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42495 (7 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

⁺A further space-filling diagram, highlighting the equatorial coordination of Hg by the six O atoms, has been deposited.



Fig. 2. Space-filling representation of [HgI₂.18C6].

macrocyclic ring, i.e. they have rotoxane-like structures. Of particular note are the closely related 18C6 complexes (Paige & Richardson, 1984; Kawasaki & Matsuura, 1984) with $CdCl_2$ and $HgCl_2$, and the DA18C6 complex (Malmsten, 1979) with HgI₂, where the gross structural features of the 1:1 complexes are analogous to that of [HgI₂.18C6]. The present structure provides another example of the relatively rare hexagonal-bipyramidal geometry. The Hg-I bond distances [2.622(1) Å] are the same as those observed by Jeffrey & Vlasse (1967) in HgI₂, but significantly shorter than the value of 2.680(1) Å reported by Malmsten (1979) for [HgI2.DA18C6]. The Hg...O distances and the I-Hg...O angles (Table 2) do not depart significantly from those reported (Paige & Richardson, 1984) for [HgCl₂.18C6]. A consequence of the accommodation of a large Hg atom at the centre of the 18C6 macrocycle is an increase in the magnitude of the mean O-C-C-O torsional angle [75 (2)°], cf. the 'normal' value of ca 65° (Goldberg, 1980).

Surprisingly, although the title complex as a whole has within statistical significance $\overline{3}$ symmetry and

adopts a very similar packing arrangement to that reported by Paige & Richardson (1984) for [HgCl₂.18C6] and [CdCl₂.18C6], the crystals are not rhombohedral. This may be a consequence of accommodating covalently larger I atoms in the structure in place of Cl atoms. In fact, the unit-cell parameters initially chosen by Paige & Richardson (1984) for a C-face-centred cell of a = 10.44, b = 11.46, c =7.75 Å, and $\beta = 82.21^{\circ}$ are all different from the present C-face-centred cell.

We thank the Science and Engineering Research Council, the Agricultural and Food Research Council, and Imperial Chemical industries PLC for financial support.

References

- ALLWOOD, B. L., CROSBY, J., PEARS, D. A., STODDART, J. F. & WILLIAMS, D. J. (1984). Angew. Chem. Int. Ed. Engl. 23, 977-979.
- CROWDER, J., HENRICK, K., MATTHEWS, R. W. & PODEJMA, B. L. (1983). J. Chem. Res. (S), pp. 82–83.
- GOLDBERG, I. (1980). The Chemistry of Functional Groups. Supplement E. The Chemistry of Ethers, Crown Ethers, Hydroxyl Groups and their Sulphur Analogues. Part 1, edited by S. PATAI, pp. 175-214. Chichester: John Wiley.
- HENRICK, K., MATTHEWS, R. W., PODEJMA, B. L. & TASKER, P. (1982). J. Chem. Soc. Chem. Commun. pp. 118-119.
- HUGHES, D. L. & TRUTER, M. R. (1983). Acta Cryst. B39, 329-336.
- International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)
- JEFFREY, G. A. & VLASSE, M. (1967). Inorg Chem. 6, 396-399.
- KAWASAKI, Y. & MATSUURA, Y. (1984). Chem. Lett. pp. 155-158.
- MALMSTEN, L. (1979). Acta Cryst. B35, 1702–1704.
- PAIGE, C. R. & RICHARDSON, M. F. (1984). Can. J. Chem. 62, 332-335.
- SHELDRICK, G. M. (1983). SHELXTL, revision 4.0, January 1983. An Integrated System for Solving, Refining and Displaying Crystal Structures from Diffraction Data. Univ. of Göttingen.

Acta Cryst. (1986). C42, 53-56

Structures of Dibromo[N,N'-bis(2-pyridylmethylene)-1,3-propanediamine]copper(II) (1) and Bromo[N,N'-bis(2-pyridylmethylene)-1,4-butanediamine]copper(II) Bromide (2)

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(Received 16 May 1985; accepted 20 September 1985)

Abstract. (1): $[CuBr_2(C_{15}H_{16}N_4)], M_r = 475.7, \text{mono$ $clinic, } C2/c, a = 13.286 (3), b = 9.723 (3), c = 13.234 (2) Å, \beta = 107.54 (1)^\circ, V = 1630.1 (5) Å^3, Z = 4, D_m = 1.94 (3), D_x = 1.94 \text{ g cm}^{-3}, \lambda(\text{Mo } K\alpha) =$ 0.7107 Å, $\mu = 65.8 \text{ cm}^{-1}$, F(000) = 932, T = 293 K, R = 0.031 for 1238 observed $[I > 2\sigma(I)]$ of 1898 unique reflections. (2): $[\text{CuBr}(\text{C}_{16}\text{H}_{18}\text{N}_4)]\text{Br}$, $M_r = 489.8$, monoclinic, $P2_1/c$, a = 12.142 (3), b =

0108-2701/86/010053-04\$01.50 © 1986 International Union of Crystallography